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# Structure of 4,4',6,6'-Tetrakis(methylthio)-2,2'-bithieno[3,4-d][1,3]dithiolylidene 

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#### Abstract

C}_{14} \mathrm{H}_{12} \mathrm{~S}_{10}, M_{r}=500 \cdot 91\), triclinic, $P \overline{1}, a=$ 7.050 (2),$\quad b=8.229$ (2), $\quad c=9.540$ (2) $\AA, \quad \alpha=$ 112.08 (2) $, \quad \beta=78.87(2), \quad \gamma=100.97(2)^{\circ}, \quad V=$ 499.0 (2) $\AA^{3}, Z=1, D_{x}=1.667 \mathrm{Mg} \mathrm{m}^{-3}, \lambda($ (Mo $K \alpha)$ $=0.71069 \AA, \mu=1.059 \mathrm{~mm}^{-1}, \quad F(000)=256, T=$ $293 \mathrm{~K}, R=0.038$ for 1984 observed reflections. The conformation of the molecular frame is that of a very shallow chair with the 1,3 -dithiole rings of envelope type. One of the exocyclic methylthio groups is perpendicular to the frame while the other is approximately coplanar to the molecular plane. The molecules form a sheet-like network nearly parallel to the ( $\overline{1} 20$ ) plane where the molecules are connected with van der Waals $S \cdots S$ contacts of 3.642 (2) and 3.753 (2) $\AA$. The intersheet interaction is also van der Waals type and unfavorable to the intermolecular charge-transfer interaction.


Introduction. Recently many efforts have been made to obtain new types of organic donors and acceptors. The title compound, (I), a thiophene-fused tetrathiafulvalene (TTF) modified by methylthio groups, has been prepared to introduce interstack interactions (Kobayashi, 1985). Unfortunately, the molecule does not show a strong electron-donating character. The crystal structure analysis was carried out to investi-

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gate relations between structure and donating character.

(I)

Experimental. Yellow needles, $0.70 \times 0.35 \times$ 0.35 mm , unit-cell parameters by least squares from 25 reflections ( $30<2 \theta<47^{\circ}$ ), Rigaku AFC-4 diffractometer with graphite monochromator, $2 \leq 2 \theta \leq$ $55^{\circ}, h=-9 \rightarrow 9, k=0 \rightarrow 10, \quad l=-12 \rightarrow 12, \omega-2 \theta$ scan, scan range $(1.4+0 \cdot 4 \tan \theta)^{\circ}$, scan speed $4^{\circ} \mathrm{min}^{-1}$ in $2 \theta$, three reflections ( $13 \overline{2}, 211, \overline{2} 40$ ) monitored every 50 reflections, variations within $3 \%$. 2591 reflections measured, 2422 unique, $R_{\text {int }}=0.02,1984$ observed, $\left|F_{o}\right| \geq 3 \sigma(F)$. No absorption corrections were applied. Structure solved by MULTAN78 (Main, Hull, Lessinger, Germain, Declercq \& Woolfson, 1978). H atoms were found from the difference map. Block-diagonal least-squares refinement with anisotropic temperature factors for non- H atoms and isotropic ones for H . $\sum w\left(\left|F_{c}\right|-k^{-1}\left|F_{o}\right|\right)^{2}$ was minimized. $w=1$ for all reflections. $\Delta \rho_{\text {max }}=0.3 \mathrm{e}^{-3}, \Delta / \sigma_{\text {max }}=0.08$ for © 1990 International Union of Crystallography

Table 1. Positional parameters $\left(\times 10^{4}\right)$ and equivalent isotropic temperature factors $\left(\AA^{2}\right)$ for non -H atoms

| $B_{\text {eq }}=(4 / 3) \sum_{i} \sum_{j} \beta_{i j} \mathbf{a}_{i} \cdot \mathbf{a}_{j}$. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $\boldsymbol{x}$ | $y$ | $z$ | $B_{\text {eq }}$ |
| S(2) | 2024 (1) | 1206 (1) | -1503 (1) | 3.15 (2) |
| S(5) | 7535 (1) | 3652 (1) | 24 (1) | 3.40 (3) |
| S(8) | 1969 (1) | 1468 (1) | 1719 (1) | $3 \cdot 68$ (3) |
| S(9) | 6452 (1) | 3022 (1) | -3108 (1) | $3 \cdot 58$ (3) |
| S(11) | 6319 (2) | 3342 (1) | 3208 (1) | $4 \cdot 26$ (3) |
| C(1) | 832 (4) | 559 (4) | 44 (4) | $2 \cdot 68$ (9) |
| C(3) | 4176 (4) | 2134 (4) | -699 (3) | 2.48 (9) |
| C(4) | 5899 (5) | 2826 (4) | -1314 (4) | 2.86 (10) |
| C(6) | 5854 (5) | 3069 (4) | 1403 (4) | 2.99 (10) |
| C(7) | 4144 (4) | 2263 (4) | 826 (3) | $2 \cdot 62$ (9) |
| C(10) | 7656 (8) | 1098 (7) | -4207 (5) | $5 \cdot 28$ (16) |
| C(12) | 8028 (7) | 5290 (7) | 3666 (5) | $5 \cdot 59$ (17) |

Table 2. Bond distances $(\AA)$ and angles $\left({ }^{\circ}\right)$

| $\mathrm{S}(2)-\mathrm{C}(1)$ | 1.764 (4) | $\mathrm{S}(9)-\mathrm{C}(10)$ | 1.801 (6) |
| :---: | :---: | :---: | :---: |
| $\mathrm{S}(2)-\mathrm{C}(3)$ | 1.744 (3) | S(11)-C(6) | 1.738 (4) |
| $\mathrm{S}(5)-\mathrm{C}(4)$ | 1.743 (4) | $\mathrm{S}(11)-\mathrm{C}(12)$ | 1.775 (6) |
| $\mathrm{S}(5)-\mathrm{C}(6)$ | 1.738 (4) | $\mathrm{C}(1)-\mathrm{C}\left(1^{1}\right)$ | 1.341 (7) |
| $\mathrm{S}(8)-\mathrm{C}(1)$ | 1.766 (4) | $\mathrm{C}(3)-\mathrm{C}(4)$ | $1 \cdot 362$ (5) |
| $\mathrm{S}(8)-\mathrm{C}(7)$ | 1.743 (4) | $\mathrm{C}(3)-\mathrm{C}(7)$ | 1.414 (5) |
| $\mathrm{S}(9)-\mathrm{C}(4)$ | 1.741 (4) | $\mathrm{C}(6)-\mathrm{C}(7)$ | $1 \cdot 369$ (5) |
| $\mathrm{C}(1)-\mathrm{S}(2)-\mathrm{C}(3)$ | 94.4 (2) | $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(7)$ | 114.1 (3) |
| $\mathrm{C}(4)-\mathrm{S}(5)-\mathrm{C}(6)$ | $92 \cdot 6$ (2) | $\mathbf{S}(5)-\mathrm{C}(4)-\mathrm{S}(9)$ | $122 \cdot 6$ (2) |
| $\mathrm{C}(1)-\mathrm{S}(8)-\mathrm{C}(7)$ | 94.2 (2) | $\mathrm{S}(5)-\mathrm{C}(4)-\mathrm{C}(3)$ | 109.8 (3) |
| $\mathrm{C}(4)-\mathrm{S}(9)-\mathrm{C}(10)$ | 101.9 (2) | $\mathbf{S}(9)-\mathrm{C}(4)-\mathrm{C}(3)$ | 127.6 (3) |
| $\mathrm{C}(6)-\mathrm{S}(11)-\mathrm{C}(12)$ | $103 \cdot 3$ (2) | $\mathrm{S}(5)-\mathrm{C}(6)-\mathrm{S}(11)$ | 125.1 (2) |
| $\mathrm{S}(2)-\mathrm{C}(1)-\mathrm{S}(8)$ | 115.9 (2) | $\mathrm{S}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | 109.9 (3) |
| $\left.\mathrm{S}(2)-\mathrm{C}(1)-\mathrm{C}(1)^{\prime}\right)$ | $122 \cdot 1$ (3) | $\mathrm{S}(11)-\mathrm{C}(6)-\mathrm{C}(7)$ | 124.8 (3) |
| $\mathrm{S}(8)-\mathrm{C}(1)-\mathrm{C}\left(1^{1}\right)$ | 122.0 (3) | $\mathbf{S}(8)-\mathbf{C}(7)-\mathrm{C}(3)$ | 117.1 (3) |
| $\mathrm{S}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | 129.1 (3) | $\mathbf{S}(8)-\mathrm{C}(7)-\mathrm{C}(6)$ | 129.3 (3) |
| $\mathrm{S}(2)-\mathrm{C}(3)-\mathrm{C}(7)$ | 116.8 (2) | $\mathrm{C}(3)-\mathrm{C}(7)-\mathrm{C}(6)$ | 113.6 (3) |

Symmetry code: (i) $-x,-y,-z$.
non-H, 0.17 for H atoms. $R=0.038, w R=0.038, S$ $=0.646$. Atomic scattering factors from International Tables for X-ray Crystallography (1974). All computations were performed on a HITAC M260D computer of the Data Processing Center of the University of Electro-Communications with the programs UNICSIII (Sakurai \& Kobayashi, 1979), MULTAN78 (Main et al., 1978) and ORTEPII (Johnson, 1976). The final atomic parameters are given in Table 1.*

Discussion. The molecular structure with the atomic numbering is shown in Fig. 1. Bond distances and angles are listed in Table 2. The molecule has a crystallographic center of inversion and the molecular frame has a very shallow chair form. The plane (I) defined by the central six atoms, $\mathrm{C}(1), \mathrm{C}\left(1^{i}\right), \mathrm{S}(2)$, $\mathbf{S}\left(2^{i}\right), \mathbf{S}(8)$, and $\left.\mathbf{S}\left(8^{i}\right)[(\mathrm{i})=-x,-y,-z)\right]$, is planar

[^1]within 0.001 (3) $\AA$. The 1,3 -dithiole ring is in an envelope conformation. The dihedral angle between plane (I) and plane (II) defined by the four atoms $S(2), C(3), C(7)$ and $S(8)$ is $14 \cdot 2(1)^{\circ}$. The dihedral angle between plane (II) and the terminal thiophene ring (III) is $1.9(1)^{\circ}$. The conformations of the two independent exocyclic methylthio groups are quite different; the $\mathrm{S}(9)-\mathrm{C}(10)$ bond is almost perpendicular to the molecular frame while $\mathrm{S}(11)-\mathrm{C}(12)$ is


Fig. 1. The molecular structure with atomic numbering. The thermal ellipsoids for non-H atoms are drawn at $50 \%$ probability.

(a)


(b)

Fig. 2. (a) The projection of the crystal structure viewed along the $c$ axis. (b) Sheet-like network of the molecules parallel to the ( 120 ) plane. Distances are given in $\AA$.
approximately coplanar to the molecular plane. The torsion angles $\mathrm{S}(5)-\mathrm{C}(4)-\mathrm{S}(9)-\mathrm{C}(10)$ and $\mathrm{S}(5)-\mathrm{C}(6)-\mathrm{S}(11)-\mathrm{C}(12)$ are 86.6 (2) and 32.7 (3) ${ }^{\circ}$, respectively.

The distances of the $S(2)-C(1)$ and $S(8)-C(1)$ bonds ( 1.764 and $1.766 \AA$, respectively) are slightly longer than those of the other endocyclic $\mathrm{S}-\mathrm{C}$ bonds ( $1.738-1.744 \AA$ ). The $\mathrm{C}(3)-\mathrm{C}(7)$ distance is longer than that of the terminal $\mathrm{C}=\mathrm{C}$ bond of TTF and the central $C=C$ distance $[1.341$ (7) $\AA$ ] is slightly shorter than that of the neutral TTF molecule ( $1.349 \AA$ ) (Cooper, Kenny, Edmonds, Nagel, Wudl \& Coppens, 1971). This dimension of the TTF moiety of this compound is very close to that of dibenzotetrathiafulvalene (Emge, Wiygul, Chappell, Bloch, Ferraris, Cowan \& Kistenmacher, 1982) which corresponds to a neutral TTF from the comparison of the structures of TTF in various electronic states (Kistenmacher, Phillips \& Cowan, 1974; Yakushi, Nishimura, Sugano, Kuroda \& Ikemoto, 1980). The exocyclic $\mathrm{S}-\mathrm{C}$ bonds, $\mathrm{S}(9)-\mathrm{C}(10)$ and $\mathrm{S}(11)-\mathrm{C}(12)$, are longer than the other $\mathrm{S}-\mathrm{C}$ bonds, $\mathbf{S}(9)-\mathbf{C}(4)$ and $\mathbf{S}(11)-\mathbf{C}(6)$. The difference of these lengths corresponds to that between $S-\mathrm{C}_{s p^{3}}$ and $\mathrm{S}-\mathrm{C}_{s p^{2}}$.

Fig. 2(a) shows the crystal structure viewed along the $c$ axis. The molecules form a sheet-like network nearly parallel to the ( $\overline{1} 20$ ) plane as shown in Fig. $2(b)$. The molecules in a sheet are connected with van der Waals $\mathrm{S} \cdots \mathrm{S}$ contacts such as $\mathrm{S}(11) \cdots \mathbf{S}\left(\mathrm{g}^{\text {ii }}\right)[(\mathrm{ii})=$ $x, y, z+1,3 \cdot 642(2) \AA]$ and $\mathrm{S}(5) \cdots \mathrm{S}\left(5^{\mathrm{iii}^{\prime}}\right)[(\mathrm{iii})=2-x$,
$1-y,-z, 3.753(2) \AA]$ and there are cavities between molecules along the $c$ axis which are filled with methylthio groups of the adjacent sheet. The interplanar separation between the thiophene plane (III) of the molecule ( $x, y, z$ ) and plane (I) of the molecule $(x+1, y, z)$ is $3 \cdot 556(8) \AA$.

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# Structures of 2,6-Bis(methylthiomethyl)phenyl Phenyl Sulfoxide and 9,18-Epithio-2,11-dithia[3.3]metacyclophane-Trichloromethane (1/1) 

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[^1]:    * Lists of structure factors, anisotropic thermal parameters and H -atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 52991 ( 14 pp .). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

[^2]:    $0.443 \mathrm{~mm}^{-1}, F(000)=680, T=293 \mathrm{~K}, R=0.044$ for 2873 observed reflections. (II) 7,15,18-Trithiatetracyclo[7.7.3.0 $\left.{ }^{5,17} 0^{13,19}\right]$ nonadeca-1 (17),2,4,9,11,13(19)-hexaene-trichloromethane (1/1), $\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{~S}_{3} . \mathrm{CHCl}_{3}$, $M_{r}=421.86$, monoclinic, $P 2_{1} / m, a=8.749$ (1), $b=$ 12.007 (2), $\quad c=8.686$ (1) $\AA, \quad \beta=92.46(1)^{\circ}, \quad V=$

